HYDROXYACIDS

B. A. Trofimov, A. G. Mal'kina, Yu. M. Skvortsov,

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L. V. Sokolyanskaya, and A. I. Gritsa

Esters of acetylenic hydroxyesters, such as type I, are easily obtained by the reactions of acetylenic alcohols with carbon monoxide [1]; they present several new opportunities for the construction of more complex heterocyclic systems.

For instance, the addition of sulfide ions to ester I in aqueous media under mild conditions (5-25°C) does not stop with the formation of the corresponding tetrasubstituted divinyl-sulfide, as reported previously [2], but rather proceeds unexpectedly to generate the novel polyfunctional heterocyclic system II.

Apparently, one of the hydroxyl groups in the intermediate divinylsulfide adds crosswise to one of the double bonds, forming a 1,3-oxathiolane ring, whereas the second hydroxyl group reacts with the ester functional group to give a lactone ring. Thus, ester I, as well as various homologs with different substituents in the 4-position, generates isomerically pure cyclization products, in which the S-C-CHCO₂Me fragment is assigned the Z-configuration, based on the fact that reaction of propiolic acid under these conditions generates exclusively Z,Z-di(2-carboxyvinyl)sulfide [3].

 $\frac{18-\text{Methoxycarbonylmethylene-7,11-dioxa-19-thiatrispiro[5.3.1.5.2]}{\text{nonadecan-8-one (II).}}{\text{was prepared from 2 mmoles of ester I and 1 mmole of Li_2S in 20 ml of H_2O (21-26°C, 4 h). Yield, 19%, mp 183°C. PMR spectrum (CCl_4), &: 1.72 (10H, m), 3.02 (2H, d), 3.75 (3H, s), 5.74 ppm (1H, s). IR spectrum: 1190, 1600, 3070, 1690, 1780 cm⁻¹.$

 $\frac{3\text{-Methoxycarbonylmethylene-2,2,6,6-tetramethyl-1,7-dioxa-4-thiaspiro[4.4]nonan-8-one.}{\text{This was prepared in an analogous manner from methyl-4-methyl-4-hydroxy-2-pentanoate; mp 157-158°C. PMR spectrum (CCl₄), <math>\delta$: 1.43 (12H, s), 3.07 (2H, d), 3.76 (3H, s), 5.76 ppm (1H, s). IR spectrum: 1190, 1590, 3050, 1690, 1780 cm⁻¹.

Both cyclization products gave satisfactory elemental analyses.

LITERATURED CITED

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